

Supporting information to Non-Boltzmann Luminescence in NaYF₄:Eu³⁺: Implications for Luminescence Thermometry

Robin G. Geitenbeek, Harold W. de Wijn, Andries Meijerink

Experimental section

Microcrystalline NaYF₄ was prepared via a solid state synthesis.³² In short, 15 mmol of NaF, 15 mmol of RE₃ (RE = Y, Eu) and 13.5 mmol of NH₄F were mixed with a pestle and mortar and afterwards placed in an alumina crucible. The mixture was fired in the oven in an excess of NH₄F under a nitrogen atmosphere. The samples were heated to 300 °C for 3 h and afterwards heated to 550 °C for 8 h. The heating rate was 5 °C min⁻¹ for both heating steps.

After synthesis, the samples were crushed with a pestle and mortar and the crystallinity was confirmed with x-ray diffraction measurements using a Philips PW1729 x-ray diffractometer. To remove the NaF impurities, samples were dispersed in H₂O and after dissolving the NaF the samples were centrifuged, decanted and dried to remove the H₂O.

The synthesis product was analyzed using scanning electron microscopy (SEM), powder X-ray diffraction (XRD) and inductively coupled plasma optical emission spectroscopy (ICP-OES). The as-synthesized particles were micrometer sized and mainly hexagonal NaYF₄ as shown in Figs. S1 and S2. The NaF impurities were successfully removed by washing with H₂O as shown in Fig. S2. The Eu³⁺ dopant concentrations, determined with ICP-OES, are 0.4%, 5.5% and 8.4% for the three different samples. The low dopant sample (0.4% Eu³⁺) is used for the main part of the work while the two higher concentration samples are used to investigate the role of cross-relaxation measurements.

Luminescence measurements were performed using a 5 mW 405 ± 10 nm diode laser and an Ocean Optics QE Pro010451 CCD. Luminescence decay measurements were measured with an Ekspla NT 342B tunable laser, a Triax 550 single emission monochromator, a Hamamatsu R928 PMT and a EG&G Ortec Turbo-MCS multichannel scaler.

Cross-relaxation

In order to expand the temperature range of luminescence thermometry with Eu³⁺ to lower temperatures, it is desirable to enhance the relaxation rate between the ⁵D₁ and ⁵D₀ levels. A potential mechanism to induce additional relaxation in this system is cross-relaxation, radiationless energy transfer between neighboring ions having resonant, or nearly resonant, transitions. A prerequisite is that the dopant concentration is sufficiently high for the dopant ions to interact.

In NaYF₄:Eu³⁺, cross-relaxation is possible since the energy of ca. 1800 cm⁻¹ released in a radiationless transition from ⁵D₁ to ⁵D₀ may be transferred to a neighboring Eu³⁺ ion residing in the ⁷F₀ ground state, resulting in a transition to the ⁷F₃ state at approximately 1800 cm⁻¹. The residual energy from the mismatch between the two different transitions is compensated by the emission or absorption of a low-energy acoustic phonon. At higher temperatures, the ⁷F₃ level becomes thermally accessible and therefore the reverse process becomes relevant as well. Additionally, the transition between the ⁷F₂ and ⁷F₄ levels involves approximately the same energy (ca. 1800 cm⁻¹) and can also contribute to the cross-relaxation processes once the ⁷F₂ and ⁷F₄ levels are thermally accessible. Note that although the ⁷F₂ and ⁷F₄ levels become accessible at a higher temperature than the ⁷F₀ and ⁷F₃ levels, the transition matrix elements connecting the ⁷F₂ and ⁷F₄ levels are larger due to the favorable ΔJ value of 2. Based on Judd-Ofelt theory the transition probabilities for $\Delta J = 0, 2, 4$ or 6 (not $0-0$) transitions are possible as forced electric dipole transition while the $\Delta J = 3$ (⁷F₀-⁷F₃) transition is not.² The reduced matrix elements $U^{(\lambda)}$ are all zero for the ⁷F₀-⁷F₃ transition, while the matrix elements are large for the ⁷F₂-⁷F₄ transition ($U^{(2)2} = 0.2226$, $U^{(4)2} = 0.0062$ and $U^{(6)2} = 0.0329$).²¹ Typically, forced electric dipole transitions ($\Delta J = 0, 2, 4$ or 6) are an order of magnitude stronger than other $J-J$ transitions which makes cross-relaxation involving the ⁷F₂-⁷F₄ transition favorable once these levels are thermally populated.

Cross-relaxation depends on the distribution of the Eu³⁺ ions over the lattice, notably the presence of nearest and next-nearest neighbors. A configuration of Eu³⁺ neighbors thus gives rise to a specific cross-relaxation rate that will vary for different Eu³⁺ ions. Faster cross-relaxation will occur for Eu³⁺

ions with one (or more) nearest Eu^{3+} neighbors. This will result in a multi-exponential decay. This is observed for samples with higher Eu^{3+} dopant levels of 5.5% and 8.4% as shown in Figs. S3 and S4. However, if the total luminescence is monitored via steady-state measurements, the temporal dependence of the various relaxation processes are averaged out and the intensity variations can be analyzed using an effective temperature-dependent cross-relaxation rate.

To explore the effect of the Eu^{3+} dopant concentration on the temperature-dependent ${}^5\text{D}_1/{}^5\text{D}_0$ intensity ratio, the luminescence of two samples with a higher Eu^{3+} dopant concentration, 5.5% and 8.4%, have been measured as shown in Fig. S5. From the spectra, the fluorescence intensity ratio of the ${}^5\text{D}_1$ and ${}^5\text{D}_0$ level has been calculated for temperatures ranging from 300–900 K and are shown in Fig. 5.

To model the cross-relaxation effects, we rewrite Eq. (12) to include

$$X_1 = g_1(X_{03}N'_0 + X_{24}N'_2) \quad (\text{S1})$$

and

$$X_2 = g_2(X_{03}N'_3 + X_{24}N'_4). \quad (\text{S2})$$

Here, X_{03} and X_{24} contain the relevant matrix elements for the transitions between the ${}^7\text{F}_0$ – ${}^7\text{F}_3$ and ${}^7\text{F}_2$ – ${}^7\text{F}_4$ respectively and the populations of the ${}^7\text{F}_J$ states (having energy E_J) of the neighboring Eu^{3+} ions, N'_J is given by

$$N'_J = N'(2J + 1) \frac{\exp(-E_J/kT)}{Z}, \quad (\text{S3})$$

in which N' measures the density of adjacent Eu^{3+} contributing to cross-relaxation and Z is the state sum. Note that we assume here that the energy of the ${}^7\text{F}_J$ levels (E_J) is not affected by interactions with the neighboring Eu^{3+} ion in the ${}^5\text{D}_0$ or ${}^5\text{D}_1$ level. The phonon processes to compensate for the energy mismatch during cross-relaxation are also assumed to be much faster than cross-relaxation and are therefore omitted from Eqs. (S1) and (S2).

To quantify the added effect of cross-relaxation in Fig. 5, the data from the 5.5% (red) and 8.4% (orange) sample were fitted using equation Eq. (12) with Eqs. (S1) and (S2). In the fitting procedure all obtained fitting parameters from the 0.4% sample (blue) were maintained, resulting in only 2 additional fitting parameters, X_{03} and X_{24} . The model with obtained values, ($X_{03} = 0.41 \pm 0.05 \text{ ms}^{-1}$ and $X_{24} = 4.0 \pm 0.4 \text{ ms}^{-1}$ for the 5.5% Eu^{3+} concentration and $X_{03} = 0.75 \pm 0.10 \text{ ms}^{-1}$ and $X_{24} = 9.7 \pm 1.4 \text{ ms}^{-1}$ for the 8.4% Eu^{3+} concentration) fits the data very well.

The ${}^7\text{F}_2$ – ${}^7\text{F}_4$ transition has a transition probability for cross-relaxation that is larger by an order of magnitude in comparison with the ${}^7\text{F}_0$ – ${}^7\text{F}_3$ transition. This is in line with the forced electric dipole character of the ${}^7\text{F}_2$ – ${}^7\text{F}_4$ transition. In fact, at elevated temperatures ${}^7\text{F}_2$ – ${}^7\text{F}_4$ is more effective, taking over from ${}^7\text{F}_0$ – ${}^7\text{F}_3$ above ca. 400 K as the dominant cross-relaxation process. We furthermore see that cross-relaxation rates increase roughly linearly with the Eu^{3+} concentration which is expected as the average number of Eu^{3+} neighbors increases linearly with Eu^{3+} concentration.

The experiments on the higher-doped Eu^{3+} samples have provided evidence that the temperature dependence of the ${}^5\text{D}_1/{}^5\text{D}_0$ intensity ratio is strongly influenced by the Eu^{3+} concentration. At higher concentrations, the temperature dependence of the ${}^5\text{D}_1/{}^5\text{D}_0$ intensity ratio can be used to probe temperatures above ca. 500 K. The useful temperature range for luminescence thermometry can thus be controlled and extended through the dopant concentration. The dopant concentration is identified as an important parameter in the performance of (nano)thermometers based on lanthanide-doped luminescent materials, and the temperature regime in which Boltzmann equilibrium is reached.

Figures and Tables

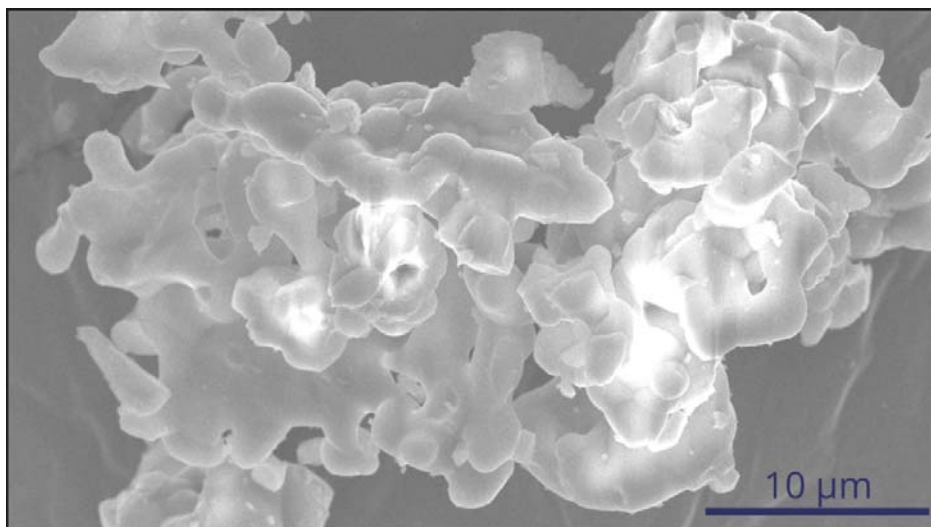


FIG. S1. SEM micrograph of the prepared $\text{NaYF}_4:\text{Eu}^{3+}$, the scalebar represents 10 μm .

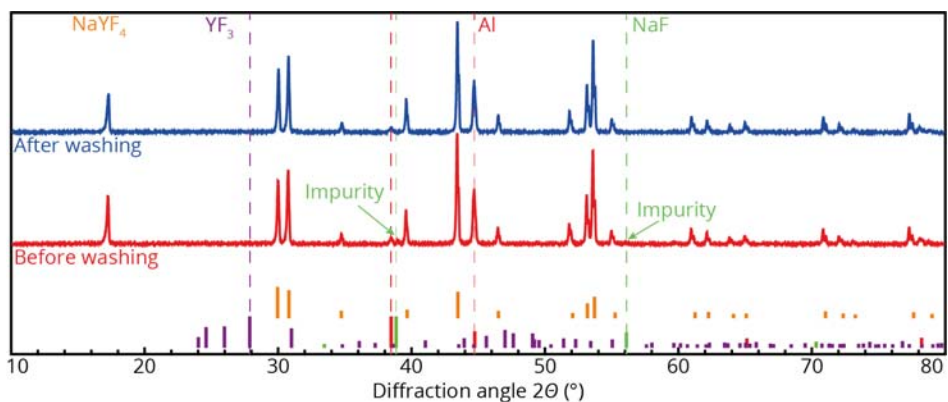


FIG. S2. X-ray diffractograms of $\text{NaYF}_4:\text{Eu}^{3+}$ before washing with H_2O to remove excess NaF (red) and after washing with H_2O (blue). The reference diffractograms include NaYF_4 (JCPDS00-064-0156, orange), YF_3 (JCPDS04-007-0883, purple), NaF (JCPDS00-036-1455, green) and aluminium from the sample holder (JCPDS00-004-0787, red). The dashed lines represent angles where the major diffraction peaks of the references are expected.

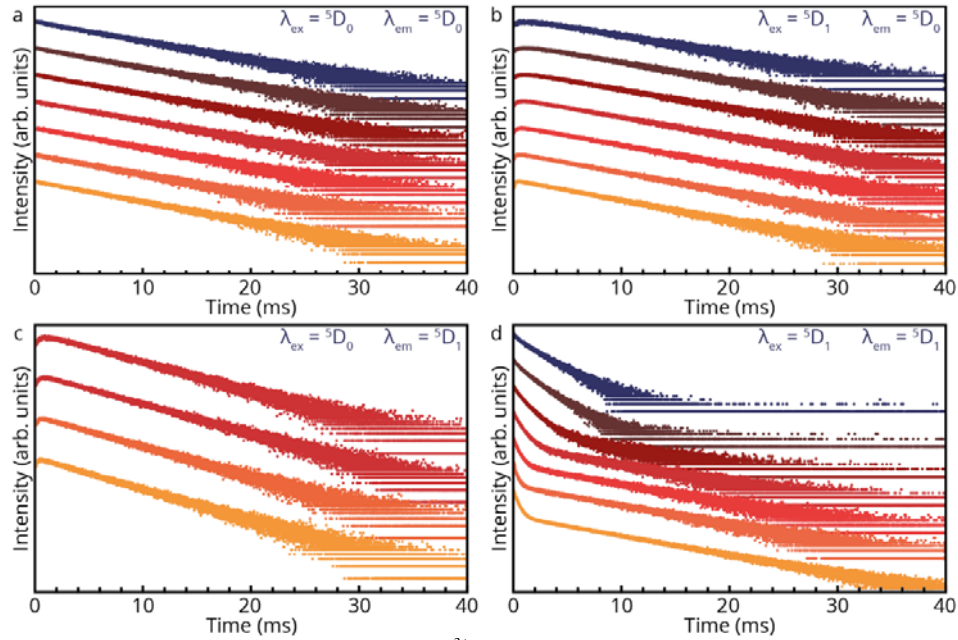


FIG. S3. Luminescence decay measurements in $\text{NaYF}_4:\text{Eu}^{3+}$ (5.5%) at temperatures ranging from 300 K (dark blue) to 900 K (orange) while exciting in the $^5\text{D}_0$ and $^5\text{D}_1$ state and monitoring the $^5\text{D}_0$ level (a and b respectively) and while exciting in the $^5\text{D}_0$ and $^5\text{D}_1$ level and monitoring the $^5\text{D}_1$ level (c and d respectively).

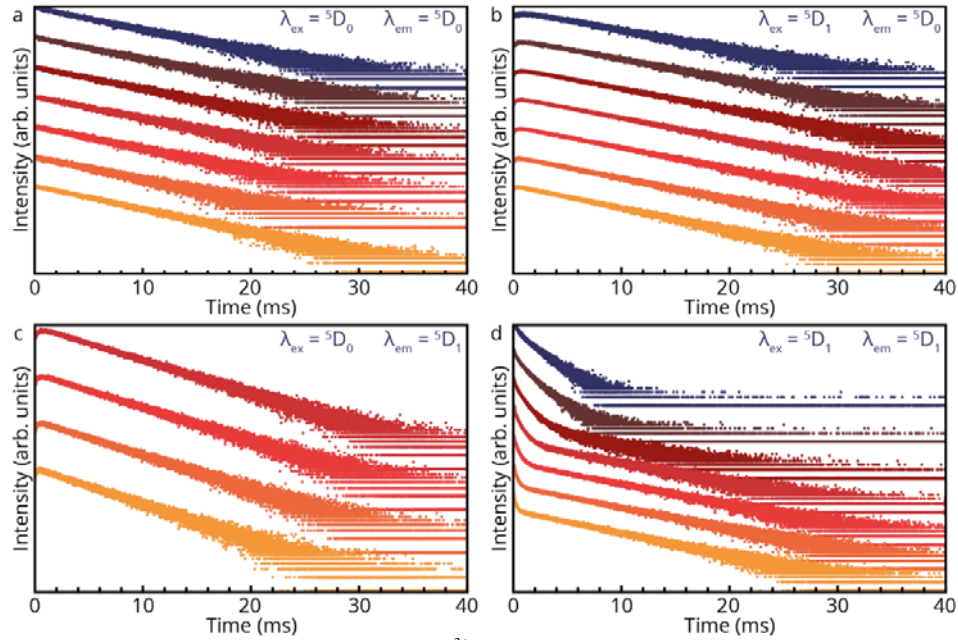


FIG. S4. Luminescence decay measurements in $\text{NaYF}_4:\text{Eu}^{3+}$ (8.4%) at temperatures ranging from 300 K (dark blue) to 900 K (orange) while exciting in the $^5\text{D}_0$ and $^5\text{D}_1$ state and monitoring the $^5\text{D}_0$ level (a and b respectively) and while exciting in the $^5\text{D}_0$ and $^5\text{D}_1$ level and monitoring the $^5\text{D}_1$ level (c and d respectively).

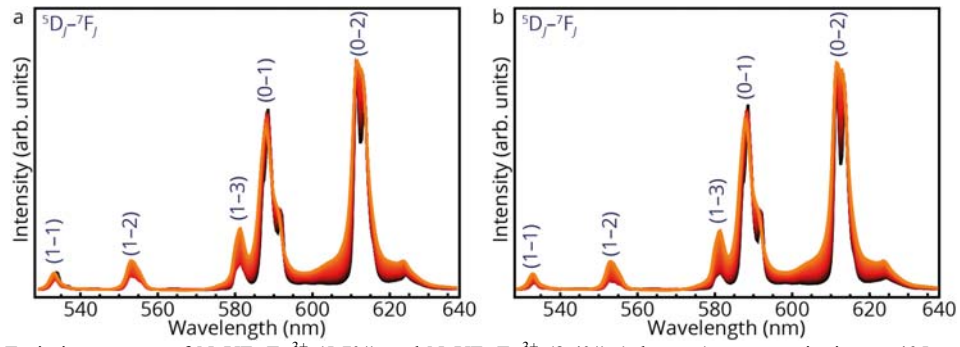


FIG. S5. Emission spectra of $\text{NaYF}_4:\text{Eu}^{3+}$ (5.7%) and $\text{NaYF}_4:\text{Eu}^{3+}$ (8.4%) (a,b resp.) upon excitation at 405 nm taken at different temperatures ranging from 300 K (black) to 900 K (orange).

TABLE SI. Fitted lifetimes (τ_1 and τ_2) of the decay measurements of the 5D_0 state upon excitation in the 5D_0 state and the corresponding radiative (τ_R) and non-radiative lifetimes (τ_{NR}).

	300 K	373 K	473 K	573 K	673 K	773 K	873 K
$1/r_1$ (ms)	6.2	6.2	6.1	6.0	5.9	5.8	5.9
$1/r_2$ (ms)	2.2	2.0	1.6	1.3	1.0	0.5	0.5
τ_R (ms)	6.2	6.2	6.1	6.0	5.9	5.8	5.9
τ_{NR} (ms)	3.4	3.0	2.3	1.7	1.2	0.5	0.6

TABLE SII. Fitted lifetimes (τ_1 and τ_2) of the decay measurements of the 5D_0 state upon excitation in the 5D_1 state and the corresponding radiative (τ_R) and non-radiative lifetimes (τ_{NR}).

	300 K	373 K	473 K	573 K	673 K	773 K	873 K
$1/r_1$ (ms)	6.3	6.4	6.2	6.1	5.9	5.8	5.9
$1/r_2$ (ms)	2.0	1.5	1.1	0.7	0.5	0.3	0.3
τ_R (ms)	6.3	6.4	6.2	6.1	5.9	5.8	5.9
τ_{NR} (ms)	3.0	2.0	1.3	0.8	0.5	0.3	0.3

TABLE SIII. Fitted lifetimes (τ_1 and τ_2) of the decay measurements of the 5D_1 state upon excitation in the 5D_0 state and the corresponding radiative (τ_R) and non-radiative lifetimes (τ_{NR}).

	300 K	373 K	473 K	573 K	673 K	773 K	873 K
$1/r_1$ (ms)	-	-	-	6.1	5.9	5.8	5.9
$1/r_2$ (ms)	-	-	-	0.7	0.5	0.3	0.3
τ_R (ms)	-	-	-	6.1	5.9	5.8	5.9
τ_{NR} (ms)	-	-	-	0.8	0.6	0.4	0.4

TABLE SIV. Fitted lifetimes (τ_1 and τ_2) of the decay measurements of the 5D_1 state upon excitation in the 5D_1 state and the corresponding radiative (τ_R) and non-radiative lifetimes (τ_{NR}).

	300 K	373 K	473 K	573 K	673 K	773 K	873 K
$1/r_1$ (ms)	6.2	6.2	6.1	6.0	5.8	5.6	5.8
$1/r_2$ (ms)	2.3	1.8	1.2	0.8	0.5	0.3	0.3
τ_R (ms)	6.2	6.2	6.1	6.0	5.8	5.6	5.8
τ_{NR} (ms)	3.7	2.5	1.5	0.9	0.6	0.4	0.4

In the fits, the consecutive photon-counting bins were weighed according to Poisson statistics, with appropriate account for low-count bins.